ME 597: FUNDAMENTALS OF ELECTROCHEMICAL ENERGY SYSTEMS Spring 2022 (Schedule: MWF 10:30 am - 11:20 am, Venue: Wang 2579)

INSTRUCTOR	Partha P. Mukherjee		
OFFICE E-MAIL	ME 2147 PHONE: 765-494-2219 pmukherjee@purdue.edu (Grades cannot be discussed via e-mail or telephone.)		
OFFICE HOURS	TBA		
GRADING POLICY	Homework/Quiz15%Examination 120%Final Examination25%Project40%		
RESOURCE	 Lectures will be developed based on the energy storage and conversion modules developed by Mukherjee and multiple resources including research papers, chapters from relevant textbooks. The following can be referred to additionally. 1. "Electrochemical Systems," Newman and Balsara (Wiley) 2. "Electrochemical Methods: Fundamentals and Applications," Bard and Faulkner (Wiley) 3. "Battery Systems Engineering," Rahn and Wang (Wiley) 4. "Handbook of Battery Materials," Eds: Daniel and Besenhard (Wiley). 5. "Modeling Transport Phenomena in Porous Media with Applications," Das, Mukherjee, Muralidhar (Springer). 6. You are free to use any analysis tools as part of this course. 		
PREREQUISITES	Strong analytical background; engineering mathematics; discussion with and approval from the instructor.		
LEARNING OUTCOMES	 The focus of the course will be on learning the fundamental concepts of energy storage and conversion with a goal to develop the ability for sound analysis. After finishing this course, the students should have the following learning outcomes. 1. Understand the fundamental physicochemical mechanisms and interactions that underlie in electrodes in an energy storage and conversion system (e.g. lithium-ion battery, polymer electrolyte fuel cell); 2. Gain knowledge of thermodynamics, kinetics, and transport phenomena (species, charge and thermal) and mechanical behavior. 3. Analyze system characteristics (performance, degradation, safety) based on the fundamental mechanisms. 		
PROJECT	The project, based on analysis of electrochemical energy systems, is an integral part of t course. You can choose your own group members (group size limited to 3). Each cours will be designed based on discussion with the instructor. The formal written report and presentation will constitute a significant part of your grade.		

Milestone	Points	Due Date
	Breakdown	
Team List (email to instructor)	-	2 nd week
Topic Selection (title and brief description of the project	5	3 rd week
objective)		
Progress Report (4-page report including the problem	10	7 th week
statement, objective, task list with tentative schedule per		
task and preliminary results)		
Draft Report	15	11 th week
Final Report and Group Presentation	70	Last day of class
Total points	100	

Project Milestone (tentative schedule)

Table of Content

(each topic: approximately 1.5-2.5 weeks)

- 1. Introduction
 - a. Basics of electrochemical energy storage & conversion (*e.g.*, lithium-ion battery and polymer electrolyte fuel cell as exemplar systems)
 - b. Operating principles & performance metrics
 - c. Current status and future perspective
- 2. Thermodynamics and Kinetics
 - a. Electrochemical concepts (open circuit potential, thermodynamic equilibrium)
 - b. Nernst Equation, Faraday's law
 - c. Butler-Volmer equation, Tafel equation
- 3. Transport Phenomena (heat, mass & momentum transfer)
 - a. Species and charge transport
 - b. Transport in electrolyte (diffusion, migration, advection)
 - c. Thermal transport
- 4. Porous Electrode
 - a. Materials (intercalation, conversion, diffusion and reaction driven)
 - b. Porous electrode theory
 - c. Interfacial phenomena
- 5. Electrochemical analysis
 - a. Circuit analysis (resistor and capacitor)
 - b. Electrochemical impedance response
 - c. Cyclic voltammetry
- 6. Performance analysis
 - a. Simplified electrode model
 - b. Coupled species and charge transport model
 - c. Thermal analysis (heat generation and thermal transport)
 - d. Mechanical analysis (diffusion induced stress)
- 7. Design consideration
 - a. Electrode & electrolyte property (electrochemical/mechanical/thermal)
 - b. Thermal safety and thermal management
 - c. Degradation (mechanical and chemical effect)
 - d. Techno-economic analysis